UNDERGROUND STORAGE TANK SECTION
FIELD/LAB FORM

NORTH CAROLINA
Dept. of Environment and Natural Resources
Division of Waste Management - UST Section

COUNTY : _____

QUAD NO: _____

LATITUDE : _____

LONGITUDE: _____

REPORT TO : **UST -** _____ Regional Office

COLLECTOR(S) : _____

DATE: _____

TIME: _____

SAMPLE PRIORITY

☐ ROUTINE ☐ EMERGENCY

☐ CHAIN OF CUSTODY

SAMPLE TYPE

☐ Soil
☐ Water
☐ Other _____

Location code:

Lab Number :
Date Received :
Time Received :
Received By :

Released By :
Date reported :

PURPOSE (circle): Baseline, Complaint, Compliance, LUST, Pesticide Study, Federal Trust, Other: _____

Field Analysis

pH _____ units Spec. Cond. at 25°C _____ umhos/cm2

Odor _____ Temperature _____ °C

Appearance _____

Field Analysis by _____

Owner: _____

Location or Site: _____

Description of sampling point: _____

Sampling Method: _____ (Pump, bailer, etc.)

Remarks: _____ (pumping time, air temp., etc)

LABORATORY ANALYSIS

	BOD		Dissolved Solids		Ag-Silver		Organochlorine Pesticides
	COD High		Fluoride		Al-Aluminum		Organophosphorus Pesticides
	COD Low		Hardness: total		As-Arsenic		Nitrogen Pesticides
	Coliform: MF Fecal		Hardness: (non-carb)		Ba-Barium		
	Coliform: MF Total		Phenols		Ca-Calcium		Acid Herbicides
	TOC		Specific Conductivity		Cd-Cadium		
	Turbitity		Sulfate		**Cr-Chromium ***		Semivolatiles
	Residue., Suspended		Sulfide		Cu- Copper		**TPH-Diesel Range**
			MBAS		Fe- Iron		
			Oil and Grease		Hg- Mercury		Volatile Organics (VOA bottle)
	pH		Silica		K-Potassium		
	Alkalinity to pH 4.5		Boron		Mg- Magnesium		**TPH-Gasoline Range**
	Alkalinity to pH 8.3		Formaldehyde		Mn-Manganese		**TPH-BTEX Gasoline Range**
	Carbonate		NH3 as N 610		Na- Sodium		
	Bicarbonate		TKN as N 625		Ni-Nickel		
	Carbon dioxide		NO2 +NO3 as N 630		**Pb-Lead ***		
	Chloride		P: Total as P		Se-Selenium		
	Chromium: Hex		PO4		Zn_Zinc		
	Color: True 80						
	Cyanide						Temperature on arrival (°C):

* Metals analyses of groundwater samples (excluding Mercury) require Standard Methods 3030C Preliminary Treatment for Acid Extractable Metals.

COMMENTS : _____